

2-PHENYL-1,2,3-OSOTRIAZOLE C-NUCLEOSIDE ANALOGS  
SYNTHESIS AND DETERMINATION OF ANOMERIC CONFIGURATION

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**Summary:** A series of anomeric 2-phenyl-1,2,3-osotriazole C-nucleoside analogs has been prepared. The anomeric configuration was determined by a novel method from the chemical shift of the C-5 proton of the osotriazole base moieties.

C-Nucleoside analogs are useful tools for biochemical investigations and for antimitotic or antiviral research<sup>1</sup>. The carbon-carbon linkage between the glycosyl and base moieties, makes these compounds more stable towards acid and enzymatic cleavage, than N-nucleosides. A satisfactory method for synthesis is by dehydration of the hydroxyalkyl chain of C-(hydroxyalkylated) nitrogen heterocyclic derivatives by use of a strong acid<sup>2-5</sup>. This method has not been extensively used because of the uncertainty of their anomeric configuration<sup>6</sup>.

In the present work 2-phenyl-1,2,3-osotriazole C-nucleoside analogs having the D-lyxo-, D-arabino-, and D-ribo-furanosyl configuration were prepared from their precursors 3,6-anhydro-heptulose phenylosazones by cyclization of the bis-(phenylhydrazone) residues with copper sulphate. Dehydration of D-galacto-heptulose phenyloszone with methanolic sulphuric acid solution<sup>8,9</sup>, then refluxing the product with copper sulphate, (Scheme 1), gave the anomeric mixture 4-( $\beta$ -D-lyxofuranosyl)-2-phenyl-1,2,3-osotriazole 1, and 4-( $\alpha$ -D-lyxofuranosyl)-2-phenyl-1,2,3-osotriazole 2, in 52% overall yield, they were separated by fractional crystallization. Compound 1; m.p. 125<sup>o</sup>,  $[\alpha]_D^{22} + 4.3^o$  (C 2.4, methanol), high resolution mass spectrum showed m/e 277.105 (calc. for C<sub>13</sub>H<sub>15</sub>N<sub>3</sub>O<sub>4</sub>; 277.106). UV (methanol)  $\lambda_{max}$  265 nm (log  $\epsilon$  4.4). Compound 2; m.p. 204<sup>o</sup>,  $[\alpha]_D^{22} + 51^o$  (C 0.4, methanol), high resolution mass spectrum showed m/e 277.106 (calc. 277.106). UV (methanol)  $\lambda_{max}$  266 nm (log  $\epsilon$  4.3). Acetylation of 1 and 2 with acetic anhydride and pyridine afforded 3 and 4 in 70% and 75% yield, respectively. Compound 3 was obtained as a syrup, high resolution mass spectrum showed m/e 403.139 (calc. for C<sub>19</sub>H<sub>21</sub>N<sub>3</sub>O<sub>7</sub>; 403.139). IR(CHCl<sub>3</sub>) 1735 cm<sup>-1</sup> (OAc). Compound 4; m.p. 114-115<sup>o</sup>, high resolution mass spectrum showed m/e 403.139 (calc. 403.139). IR (KBr) 1740 cm<sup>-1</sup> (OAc).

D-glucio-Heptulose phenylosazone afforded the anomeric mixture, 4-( $\beta$ -D-arabinofuranosyl)-2-phenyl-1,2,3-osotriazole **5** and 4-( $\alpha$ -D-arabinofuranosyl)-2-phenyl-1,2,3-osotriazole **6**, in 46% overall yield. Compound **5**; m.p. 130°,  $[\alpha]_D^{22}$  -56.5° (C 1.1, methanol), high resolution mass spectrum showed m/e 277.107 (calc. for  $C_{13}H_{15}N_3O_4$ ; 277.106). Compound **7**; m.p. 144°,  $[\alpha]_D^{22}$  +66.2° (C 0.13, methanol), high resolution mass spectrum showed m/e 277.107 (calc. 277.106).

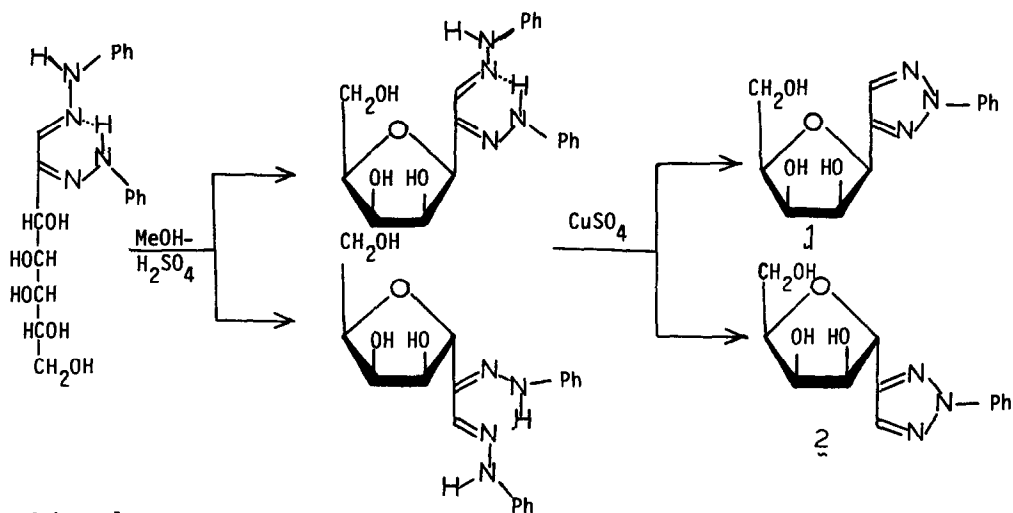
The anomeric configuration was determined by a novel method from the n.m.r. spectra of these compounds. The chemical shift of the C-5 proton of the osotriazole ring in  $\beta$ -nucleosides showed a down field shift, compared to the corresponding  $\alpha$ -nucleosides (Table 1). This deshielding effect can be explained by the proximity of the C-5 proton to the furanosyl ring oxygen in the  $\beta$ -anomer. The assignment of the anomeric configuration by this method is more advantageous than that from the chemical shift of the anomeric proton<sup>10,11</sup> since the C-5 proton can be easily distinguished as a sharp, well resolved, peak at the aromatic region. Additionally the anomeric proton of these nucleosides was exceptional; the trans  $\hat{1},\hat{2}$  proton resonance was found down field in place of the expected cis  $\hat{1},\hat{2}$  proton<sup>12,13</sup>. This is in accord with the values of the coupling constant  $J_{\hat{1},\hat{2}}$  of the anomeric proton (Table 1). Low coupling constants (less than 3.5 Hz)<sup>14</sup> can be assigned to the trans  $\hat{1},\hat{2}$  anomeric protons.

Table 1

Compd.	Chemical Shift		Coupling Constant
	H- $\hat{1}$	H- $\hat{5}$	
<b>1</b>	4.64	8.04	9.5
<b>2</b>	4.73	7.94	1.1
<b>3</b>	5.02	7.83	8.4
<b>4</b>	4.92	7.76	1.0
<b>5</b>	4.28	8.04	9.2
<b>6</b>	4.94	7.93	0
<b>7a</b>	4.60	8.04	-
<b>7b</b>	5.20	8.04	0

n.m.r. spectra at 80 MHz, in dimethyl sulphoxide-d<sub>6</sub> with CD<sub>3</sub>COOD added, for compounds **1,2,5,6** and **7a**; compounds **3,4** and **7b** chloroform-d. Internal reference tetramethylsilane.

The n.m.r. spectra were supported by the optical properties of these C-nucleosides. They obeyed Hudson isorotation rule<sup>15</sup> at the sodium D-line and were consistent with the o.r.d. results in the long wavelength region (Figure 1).



Scheme 1

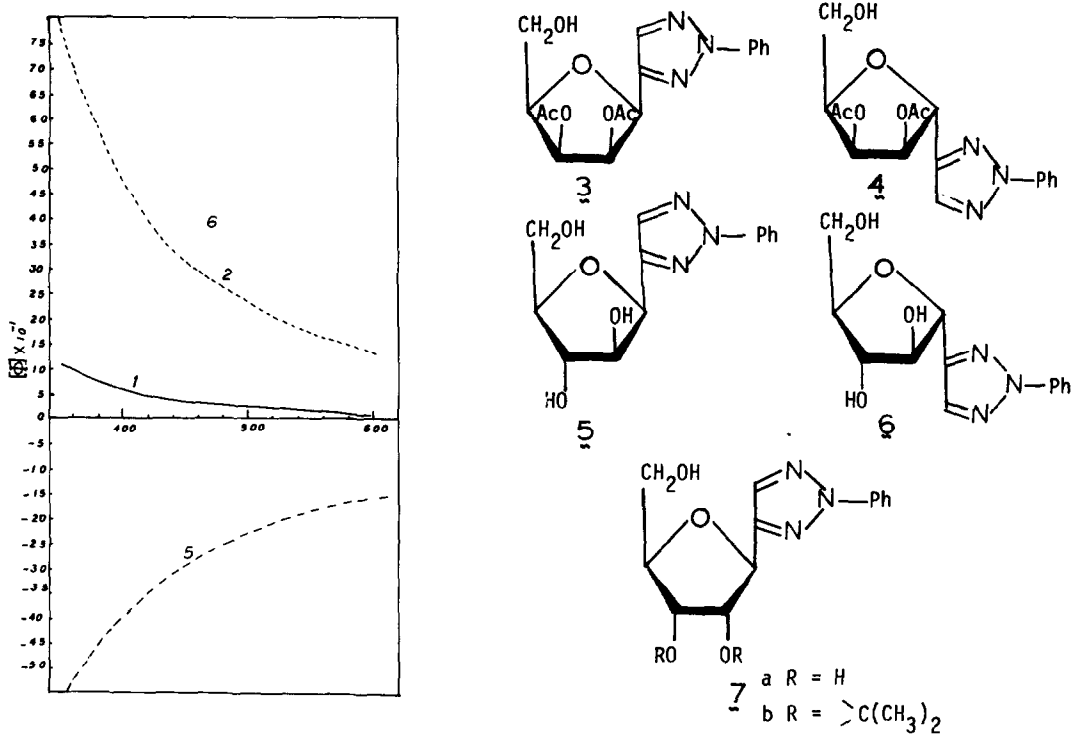


Figure 1. ORD curves in methanol of:

- (—) 4-(β-D-lyxofuranosyl) 2-phenyl-1,2,3-osotriazole (1),
- (---) 4-(α-D-lyxofuranosyl) 2-phenyl-1,2,3-osotriazole (2),
- (.....) 4-(β-D-arabinofuranosyl) 2-phenyl-1,2,3-osotriazole (5),
- (-.-.-) 4-(α-D-arabinofuranosyl) 2-phenyl-1,2,3-osotriazole (6).

4-( $\beta$ -D-Ribofuranosyl)-2-phenyl-1,2,3-oxotriazole  $\lambda$ , was obtained by similar treatments of D-altrio-heptulose phenylsazone and was assigned the  $\beta$ -D-configuration from the chemical shift of the C-5 proton. Its isopropylidene derivative  $\lambda_b$  had a low coupling constant for the anomeric proton (Table 1), the difference,  $\Delta\delta$ , between the chemical shifts of the two methyl signals (0.20 ppm) confirms the  $\beta$ -D-configuration<sup>16-18</sup>, ( $\Delta\delta$  for  $\beta$ -nucleosides is  $\geq 0.18$ , and for  $\alpha$  anomers is  $\leq 0.10$ ).

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